

## Modern Chemistry Chapter 10

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KS4 Students in Years 10 and 11 are following the new Edexcel linear GCSE specification (9-1) with first teaching in September 2015 Modern chemistry chapter 10 homework 10 1 answers. Students' progress is closely monitored by teacher's assessments of both class and homework and also by using internal school tests during the academic year.

~~Modern Chemistry Chapter 10 Homework 10 1 Answers~~

These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

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Modern Chemistry Chapter 10 States of Matter Kinetic-Molecular Theory The kinetic-molecular theory of matter is based on the idea that particles of matter are always in motion and this motion has consequences that affect its physical properties.

~~CHAPTER 10 States Matter~~

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low density, lots of empty space between particles; volume of space that gas particles take up, can be reduced by compressing particles together (assumption 1), fluid, spreads out to shape of container, no definite shape or volume (assumption 3 and 4)

~~Assessment Chapter Test A~~

Using Figure 14, estimate the approximate equilibrium vapor pressure of each of the following at the specified temperature. a. water at 80C b. diethyl ether at 20C c. ethanol at 60C

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~~Using Figure 14, estimate the approximate equilibrium ...~~

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each

statement or best answers each question. \_\_\_\_\_ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

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